COMMUNICATIONS TO THE EDITOR

THE USE OF SODIUM IN THE PREPARATION OF METHYLENE-LINKED SILANES

Sir:

The use of sodium in the preparation of compounds of the type $ASi(CH_3)_2[CH_2Si(CH_3)_2]_nB$ has been investigated. These compounds have been made wherein the A and/or B may be methyl, chloride, or ethoxy. The compounds where both A and B are methyl groups have been tentatively called "silahydrocarbons."

Compounds of the above type where A and B are methyl groups may be prepared from reactions of materials such as (CH₃)₃SiCl, CH₃[(CH₃)₂- $SiCH_2$ Cl and sodium. This reaction gives products of the type formula (CH₃)₃Si[CH₂Si- $(CH_3)_2]_n CH_3$. Yields of these reactions are 65-70% of the desired product. Two of these compounds also have been prepared by reacting (CH₃)₃SiCl and (CH₃)₂SiCl₂ with (CH₃)₃SiCH₂-MgCl.¹

The preparation of compounds where both A and B are ethoxy groups may be carried out by (1) and (2) as follows.

 $\begin{array}{c} (CH_3)_2Si(OC_2H_5)_2 + CICH_2Si(CH_3)_2OC_2H_5 + \\ 2Na \longrightarrow C_2H_5OSi(CH_3)_2[CH_2Si(CH_3)_2]_0Oc_2H_5 \end{array}$ (1)

 $NaCl + NaOC_2H_5$

 $(CH_3)_2Si(OC_2H_5)_2 + ClCH_2Si(CH_3)_2Cl + 2Na -$ (2) $C_2H_5OSi(CH_3)_2[CH_2Si(CH_3)_2]_1OC_2H_5 + 2NaCl$

In (2) it appears that $NaOC_2H_5$ is formed and reacts with the $ClSi(CH_3)_2SiCH_2(CH_3)_2OC_2H_5$ to give $C_2H_5OSi(CH_3)_2CH_2Si(CH_3)_2OC_2H_5$.

The preparation of compounds where A is a methyl group and B is ethoxy is easily effected by a reaction such as

 $(CH_3)_3SiCl + ClCH_2Si(CH_3)_2OC_2H_5 + 2Na (CH_3)_3SiCH_2Si(CH_3)_2OC_2H_5 + 2NaCl$

This reaction gives yields of 75-80% of the desired product and smaller amounts of higher members. The subsequent preparation of higher members of this series is best achieved by conversion of the ethoxy group to the chloride by reaction with acetyl chloride or benzoyl chloride and the reaction of the chloride with CICH₂Si(CH₃)₂OC₂H₅ and sodium. In this way materials of the type $(CH_3)_3Si[CH_2Si(CH_3)_2]_nOC_2H_5$ may be produced.

The reaction of either the ethoxy materials or the chloride-ended materials with Grignard reagents will give the corresponding "silahydrocarbons." The monofunctional ethoxy and chloride compounds also may be hydrolyzed to give materials of the type $[(CH_3)_3Si[CH_2Si(CH_3)_2]_n]_2O$ while hydrolysis of the difunctional compounds gives cyclic materials with the type formula $[[-(CH_3)_2SiCH_2]_nSi(CH_3)_2O-]_m.$ Two cyclic materials with formulas [(CH₃)₂SiCH₂Si(CH₃)₂-

(1) Whitmore, Sommer, Goldberg and Gold, THIS JOURNAL, 69, 980 (1947).

 O_{2} and $(CH_{3})_{2}Si[CH_{2}Si(CH_{3})_{2}]_{2}O$ have been isolated and identified.

A typical experiment in this series of reactions involves the following procedure: 800 g. of toluene and 4 moles of sodium were placed in a flask and heated to 110° with vigorous stirring. A mixture of 2 moles of (CH₃)₃SiCl and 2 moles of $ClCH_2Si(CH_3)_2OC_2H_5$ was added at such a rate that the temperature was maintained at 110°. The materials were then filtered and distilled. Distillation gave 1.57 moles of (CH₃)₃Si[CH₂Si- $(CH_3)_2]_1OC_2H_5$, b. p. 160° at 740 mm., $n^{25}D$ 1.4148, d^{25} 0.8060, a yield of 78.5%. Anal. Calcd. for $C_8Si_2H_{22}O$: Si, 29.5; C, 50.55; MRD, 59.15. Found: Si, 29.65; C, 50.6; MRD 58.9. Similar experiments have produced the compounds indicated up to n values of 4.

There is a good reason for using $ClCH_2Si-(CH_3)_2OC_2H_5$ rather than $ClCH_2Si(CH_3)_2Cl$ in these reactions. The reaction of $ClCH_2Si(CH_3)_2$ -Cl with other molecules like itself is so rapid that the simpler members of the series are very difficult to isolate. The products of the reaction of several molecules of CICH₂Si(CH₃)₂Cl and sodium are materials with high molecular weights and with the basic unit $[Si(CH_3)_2CH_2]_n$ predominating in the molecule.

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A NEW SYNTHESIS OF 1-(2',6',6'-TRIMETHYL-CYCLOHEXEN-1'-YL)-3-METHYLHEXA-1,3,5-TRIEN-7-ONE (C18 KETONE)

Sir:

The C_{18} ketone $(II)^{\scriptscriptstyle 1}$ is an important intermediate in the synthesis of "vitamin A acid" and vitamin A itself. Recently we have synthe-



(1) (a) Arens and van Dorp, Nature, 157, 190 (1946); Rec. trav. chim. 65, 338 (1946); (b) Heilbron, Jones and O'Sullivan, Nature, 157, 485 (1946); J. Chem. Soc., 868 (1946); (c) Karrer, Jucker and Schick, Helv. Chim. Acta, 29, 704 (1946).